

MR566A and MR566B, New Melanin Synthesis Inhibitors Produced by *Trichoderma harzianum*

II. Physico-chemical Properties and Structural Elucidation

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(Received for publication December 6, 1996)

New melanin synthesis inhibitors (MR566A and B) and six related known isocyanocyclopentenes were isolated from the fermentation broth of *Trichoderma harzianum*, and their structures were elucidated by spectroscopic methods. The structures of novel isocyanides, MR566A (1) and B (2), were elucidated as 1-(3-chloro-1,2-dihydroxy-4-isocyano-4-cyclopenten-1-yl)ethanol, 1-(1,2,3-trihydroxy-3-isocyano-4-cyclopenten-1-yl)ethanol, respectively. The structure of novel oxazole, MR93B (9), was elucidated as 4-[(1Z)-3-hydroxy-2-hydroxymethyl-1-propen-1-yl]oxazole.

MR566A (1) and B (2) were isolated as new melanin biosynthesis inhibitors from the culture broth of *Trichoderma harzianum* which had been isolated from a soil sample, together with a new oxazole, MR93B (9) and six known isocyanide compounds¹⁾. The eight isocyanide compounds (1~8) showed melanogenesis inhibitory activities in *Streptomyces bikiniensis* and B16 melanoma cells. In the preceding paper¹⁾ we described the taxonomy of the producing strain, fermentation, isolation and the biological activities of these compounds. In what follows, we present the elucidation of the structures of 1, 2 and 9 with relative stereochemistry of 7 and NMR assignments of 3 and 5. Compounds 3 and

5 have been purified from *T. hamatum* as rhodium complexes by BOYD *et al.*²⁾. However there were no complete NMR assignments of intact compounds.

Results and Discussion

Structural Determination of MR566A (1)

The physico-chemical properties of MR566A (1) are summarized in Table I. The observation of characteristic absorption at 2121 cm⁻¹ in the IR spectrum of 1 indicated the presence of an isocyano group³⁾. From the observation of CI-MS, the molecular weight of 1 could be assigned as 203. The molecular formula of 1 was

Fig. 1. Structures of MR566A (1), B (2), MR93B (9) and related compounds.

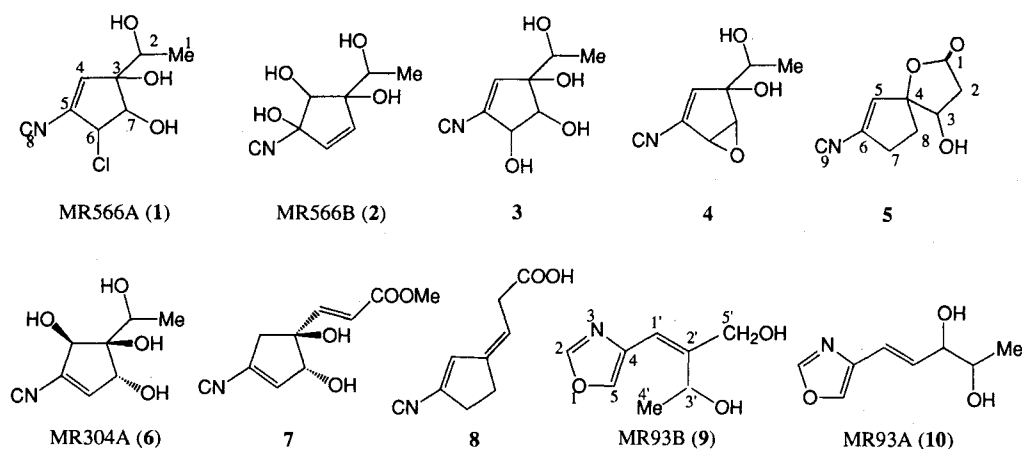


Table 1. Physico-chemical properties of 1, 2 and 9.

	1	2	9
Appearance	Brown powder	Brown powder	Colorless oil
$[\alpha]_D^{25}$	10° (c 0.1, MeOH)	70° (c 0.2, MeOH)	- 4.0° (c 0.1, MeOH)
Molecular formula	C ₈ H ₁₀ ClNO ₃	C ₈ H ₁₁ NO ₄	C ₈ H ₁₁ NO ₃
HRCI-MS <i>m/z</i> (M+H) ⁺	204.0437 (calcd 204.0427)		170.0817 (calcd 170.0812)
IR (KBr) cm ⁻¹	3361, 2121, 1051, 759	3413, 2138, 1335, 1159, 970	3361, 2929, 1681, 1517, 1400, 1068, 617
UV (MeOH)λ _{max} nm (ε)	End	End	230 (15138)

Table 2. ¹H NMR data for MR566A (1), MR566B (2), 3, 4 and 5 in CD₃OD.

Position	1 (600 MHz)	2 (300 MHz)	3 (300 MHz)	4 (300 MHz)	5 ^a (500MHz)
1	1.18 (3H, d, 6.6)	1.33 (3H, d, 6.3)	1.17 (3H, d, 6.0)	1.19 (3H, d, 6.3)	
2	3.73 (1H, q, 6.6)	3.26 (1H, q, 6.3)	3.70 (1H, q, 6.0)	3.70 (1H, q, 6.3)	2.66 (1H, dd, 18.1, 2.8) 3.16 (1H, dd, 18.1, 5.9)
3					4.5 (1H, dd, 5.9, 2.8)
4	6.18 (1H, d, 1.2)	4.12 (1H, dd, 1.2, 1.2)	6.0 (1H, d, 1.2)	5.90 (1H, dd, 2.4, 2.4)	
5					6.23 (1H, s)
6	4.72 (1H, dd, 4.5, 1.2)	6.08 (1H, dd, 6.0, 1.2)	4.49 (1H, dd, 4.5, 1.2)	3.90 (1H, dd, 2.7, 2.7)	
7	4.25 (1H, d, 4.5)	6.39 (1H, dd, 6.0, 1.2)	3.89 (1H, d, 4.5)	3.74 (1H, dd, 2.7, 2.4)	2.84 (1H, m) 2.74 (1H, m)
8					2.37 (1H, ddd, 4.5, 8.6, 14.4) 2.23 (1H, ddd, 4.5, 8.6, 14.4)

^a Sample was dissolved in D₂OTable 3. ¹³C NMR data for MR566A (1), MR566B (2), 3, 4 and 5 in CD₃OD.

Position	1 (500 MHz)	2 (500 MHz)	3 (500 MHz)	4 (300 MHz)	5 ^a (500 MHz)
1	17.7	16.8	17.9	18.8	179.3
2	71.4	75.5	71.6	72.5	38.7
3	83.2	85.1	81.9	85.1	74.5
4	137.2	87.9	134.1	138.4	101.9
5	137.1	93.0	^b	135.5	128.5
6	67.5	129.9	81.0	60.3	^b
7	79.9	142.2	79.0	58.0	33.6
8	170.9	164.6	169.7	169.0	33.5
9					169.1

^a Sample was dissolved in D₂O^b not detected

Fig. 2. NMR data and partial structures of MR566A (1), 3 and 4.

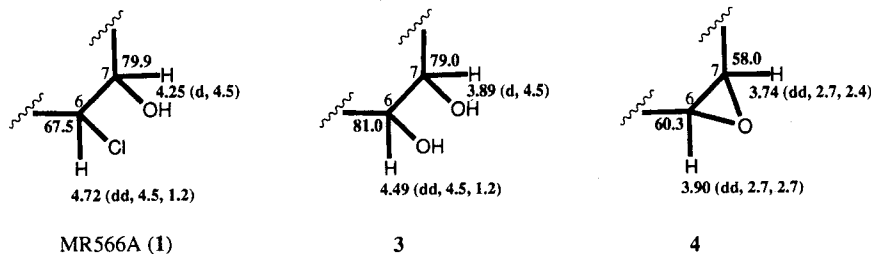


Fig. 3. Long-range C-H correlations observed in HMBC spectrum of MR566A (1).

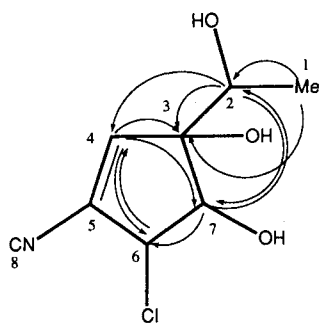
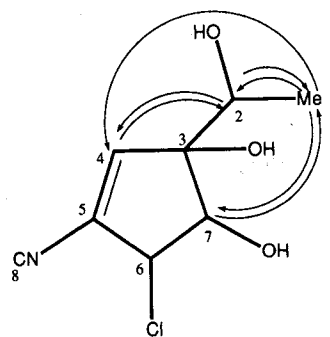


Fig. 4. NOE data of MR566A (1).



confirmed as $C_8H_{10}ClNO_3$ by the high resolution MS data [calcd.: 204.0427 for $(M+H)^+$, found: 204.0437 in the positive ion mode and calcd.: 203.0349 for M^- , found: 203.0346 in the negative ion mode]. The 1H and ^{13}C NMR spectral data for **1** are shown in Tables 2 and 3. One tri-substituted double bond, two oxygenated methines, one chlorinated methine, one oxygenated quaternary carbon, one methyl, and one isocyano carbon at 171.0 ppm were found from the 1H and ^{13}C NMR spectra of **1**. These data suggested that **1** was an analogue of the trichoviridin type isocyanide antibiotic^{4,5}. The presence of a chlorine atom was confirmed by the HR-MS data and the detection of the $M+2$ peak of the isotope having approximately one-third of the intensity of the molecular ion peak, together with absorption at 795 cm^{-1} in the IR spectrum. The position of the chlorine atom was determined to be at C-6 based on its chemical shift. In the ^{13}C NMR data of **1**, the C-6 signal was observed at 67.5 ppm which was upfield compared with that of **3** with hydroxyl group and was lower than that of **4** with epoxy (Fig. 2). The 1H NMR spectral data of **1** was similar to those of **3** and **4**, but the C-6 methine signal appeared down field compared with those of **3** and **4**. These data indicated that **1** was a chlorinated analogue of **3**. Finally the planar structure of **1** was

determined by a HMBC experiment and its data are summarized in Fig. 3. NOE data (Fig. 4) also supported the proposed structure. Based on this spectral evidence, the structure of **1** was elucidated to be 1-(3-chloro-1,2-dihydroxy-4-isocyano-4-cyclopenten-1-yl)-ethanol. According to the best of our knowledge, this is the first report of isocyanide compound with a chlorine atom from *Trichoderma* sp.

Structural Determination of MR566B (2)

The physico-chemical properties of **2** are summarized in Table 1. No parent ion for **2** was detected in several MS experiments. The molecular formula of **2** was determined to be $C_8H_{11}NO_4$ by CI-MS data of dehydrated ion peak at m/z 168 $(M+H-H_2O)^+$ and 1H and ^{13}C NMR data (Tables 2 and 3). In the ^{13}C NMR spectrum of **2**, two olefinic methines, two oxygenated methines, one methyl, and three quaternary carbons were detected. In the IR spectrum of **2**, absorption of isocyano group was observed at 2138 cm^{-1} . The carbon signal of isocyanide was observed at 164.6 ppm which is shifted upfield compared to those of α,β -unsaturated isocyano carbons (Table 3). Two sp^3 quaternary carbons were assigned to be oxygenated from chemical shifts at 85.1 ppm (C-3) and 93.0 ppm (C-5). We speculated that the extremely low chemical shift of C-5 was caused by

the attachment of the isocyano group and oxygen. All of the oxygens of the four oxygenated carbons were assigned to the carbons with a hydroxyl group. The possibility of the presence of an epoxide group was disproved by the ^{13}C chemical shift and the values of $^1J_{\text{CH}}$ for C-2 and C-4, 148 Hz and 158 Hz, respectively^{6,7}. The positions of the double bond and oxygenated carbons together with all NMR assignments were determined by C-H long-range correlations observed in the HMBC spectrum (Fig. 5). Based on these spectral data, the structure of **2** was determined to be 1-(1,2,3-trihydroxy-3-isocyano-4-cyclopenten-1-yl)ethanol. Presence of the tertiary alcohol at C-5 with an isocyano group might be the reason for the difficulty in detecting the parent ion in several MS spectra.

Relative Stereochemistry of **7**

Compound **7** has been isolated from *T. hamatum* by BALDWIN *et al.*⁸. However, there have been no reports on the stereochemistry of **7**. The relative stereochemistry of **7** was determined by NOE (Fig. 6). From the observed

NOE data between H-8 and H-5 β , the relative stereochemistry of H-8 and H-5 β could be determined to *syn*. From the strong NOE between C-3 methine proton and H-5 α , the relative configuration between C-3 and H-5 α could be proposed.

Structural Determination of MR93B (**9**)

The physico-chemical properties of MR93B (**9**) are also summarized in Table 1. Compound **9** was purified as a colorless oil. The HRCI-MS gave a (M+H)⁺ peak at m/z 170.0817 for $\text{C}_8\text{H}_{11}\text{NO}_3$ (calcd 170.0812). In the UV spectrum of **9** only a single maximum was observed at 230 nm (ϵ 15138). The ^1H , ^{13}C and HMBC NMR data of **9** are shown in Table 4. The ^1H NMR spectrum exhibited three singlet signals for H-2, H-5 and H-1', one doublet for methyl protons H-4', one doublet of doublets for methylene protons for H-5', and a quartet for methine proton H-3'. The ^{13}C NMR, C-H COSY and DEPT spectra exhibited signals for four CH carbons, one CH_2 group, one CH_3 group, and two quaternary carbons. By comparison of spectral data of **9** to those of MR93A

Fig. 5. Long-range C-H correlations observed in HMBC spectrum of MR566B (**2**).

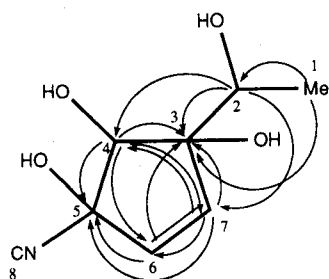


Fig. 6. NOE data with percentage of enhancement of **7**.

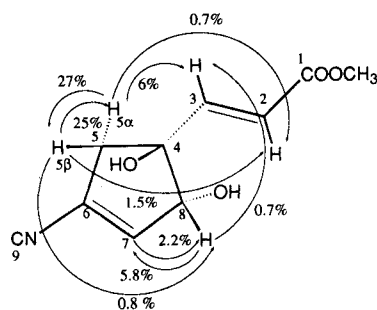


Table 4. ^1H , ^{13}C and HMBC NMR data for MR93B (**9**) in CDCl_3 .

Position	^1H (ppm)	^{13}C (ppm)	HMBC ^a
2	7.91 (1H, s)	150.7 (dd, 230.7, 7.9)	C-5
4		137.2 (s)	
5	7.66 (1H, s)	137.0 (d, 208.1)	C-4
1'	6.34 (1H, s)	112.9 (d, 149.5)	C-4, C-2', C-3', C-5'
2'		146.8 (s)	
3'	4.81 (1H, q, 6.75)	66.8 (d, 159.9)	
4'	1.44 (3H, d, 6.75)	21.9 (q, 126.4)	C-2', C-3'
5'	4.30 (2H, dd, 13.7, 48.3)	65.6 (t, 159.9)	C-1', C-2'

^a Carbon resonances that were long-range correlated with protons.

Table 5. ^{13}C NMR chemical shift and $^1J_{\text{CH}}$ of oxazole (CDCl_3).

	^{13}C (ppm)			$^1J_{\text{CH}}$ (Hz)		
	C-2	C-4	C-5	C(2)-H(2)	C(4)-H(4)	C(5)-H(5)
Oxazole*	150.6	124.5	138.1	230	194	210
9	150.7		137.0	230.7		208.1
10	151.3		137.8	230.5		206.0

* Reference (ADAMCZESKI *et al.*)¹²⁾

(**10**) and melanoxazol^{9,10)}, it was speculated that **9** was analogue of monosubstituted oxazole, **10**. The ^{13}C NMR chemical shift of C-5' (65.6 ppm) exhibited that C-5' of **9** was a hydroxyl methyl instead of an aldehyde (192 ppm) of melanoxazol. From these data the side chain was determined to be 3-hydroxy-2-hydroxymethyl-1-propenyl group. The position of side chain on the monosubstituted oxazole was established as C-4 by chemical shifts of ^{13}C NMR and HMBC data. To confirm the substitution pattern of the oxazole, the $^1J_{\text{CH}}$ values were measured. From the obtained $^1J_{\text{CH}}$ values, $^1J_{\text{C}_2\text{H}_2} = 230.7$ Hz and $^1J_{\text{C}_5\text{H}_5} = 208.1$ Hz, the position of the side chain was assigned to C-4^{9,11)} (Table 5). An *Z*-orientation of the 1',2'-double bond was established from NOE between 1'-H and 5'-H (data not shown). Based on the above mentioned spectral data, the structure of **9** was determined to be 4-[(1*Z*)-3-hydroxy-2-hydroxymethyl-1-propen-1-yl]oxazole.

Experimental

General Procedure

NMR spectra were recorded on a Bruker AMX-500 and JEOL JNM-A600 spectrometers in CD_3OD , D_2O or CDCl_3 solutions. HRCI-MS were obtained on JMS-SX102A double focusing mass spectrometer, using methane as a reagent gas. The instrument was calibrated for both negative and positive ion mode using perfluorokerosene as a reference compound. Samples were introduced into the ion source using direct insertion probe. The source temperature was set at 120°C. ESI-MS were obtained on VQ Quattro 4000. Infrared spectra (KBr pellet) and ultraviolet spectra were taken on a Laser Precision Analytical IFX-65s and Shimadzu UV-260 spectrophotometer, respectively. Optical rotation was recorded on a Schmidt+Haensch POLARTRONIC

polarimeter.

1-(1,4,5-Trihydroxy-3-isocyanocyclopenten-2-enyl)-ethanol (**3**): White powder; UV (MeOH) λ_{max} nm (ϵ) end; $\text{C}_8\text{H}_{11}\text{NO}_4$; ESI-MS m/z 184 ($\text{M} - \text{H}$)⁻.

2-Hydroxy-4-isocyano- α -methyl-6-oxabicyclo[3.1.0]hex-3-ene-2-methanol (**4**): Brown powder; UV (MeOH) λ_{max} nm (ϵ) 218 (5172); $\text{C}_8\text{H}_9\text{NO}_3$; $[\alpha]_{\text{D}} -82^\circ$, c 0.1.

4-Hydroxy-8-isocyano-1-oxaspiro[4.4]cyclonon-8-en-2-one (**5**): Brown powder; UV (MeOH) λ_{max} nm (ϵ) 229 (4453); $\text{C}_9\text{H}_9\text{NO}_3$; ESI-MS m/z 202 ($\text{M} + \text{Na}$)⁺.

Methyl-3-(1,5-dihydroxy-3-isocyanocyclopent-3-enyl)prop-2-enoate (**7**): Brown powder; UV (MeOH) λ_{max} nm (ϵ) end; $\text{C}_{10}\text{H}_{11}\text{NO}_4$; ESI-MS m/z 232 ($\text{M} + \text{Na}$)⁺.

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